Year 9: Combined Science

Curriculum Intent: Year 9 science looks to build on the foundations from years 7 and 8 and work towards developing knowledge of similar areas as students begin to work towards their GCSE examinations:

- Physics: Energy, Motion, Waves and Radiation.
- Chemistry: Particles, Periodic Table, Chemical Reactions and Earth and Environmental Science, with the addition of more complex ideas in Monitoring Reactions.
- Biology: Cell Level Systems, Scaling up, Community Level Systems, Genes, Inheritance and Selection

The subject and procedural knowledge demands increase in this year, and students will be exposed to more complex ideas, models and explanations. Practical Activities (PAG) will be coupled with mathematical skills to build more procedural knowledge, focusing on areas such as data analysis and interpretation of graphs. The scientific knowledge developed through this year will provide the basis for further study in these areas as students enter year 10, with the whole of the Combined Science content being taught before the year 10 PPE's.



	year 10 PPE's.					
	Autumn 1	Autumn 2	Spring 1	Spring 2	Summer 1	Summer 2
	Chemistry module	Physics module P1	Physics module P2 -	Physics module 4-	Chemistry module 2	End of Year exams and
	C1 - particles	-Matter	forces	waves	 Relative formula 	review
	Particles	Density	• Speed	 EM waves and EM 	mass	
	Chemical &	Gas pressure &	 Vectors and scalars 	spectrum	 Empirical formula 	Physics module- P3
	physical changes	temperature	 Acceleration 	Isotopes	Filtration and	electricity
	 Atomic structure 		 Distance- time graph 	 Alpha, beta and 	crystallisation	 Introduction to
	History of the	Chemistry module	 Velocity – time graphs 	gamma radiation	 Distillation 	circuits and basic
	development of	C2 – elements &	 Kinetic energy and 	 Nuclear equations 	 Chromatography 	circuit rules
	the atom	compounds	motion calculations	Half life		Dissiplinary by souled as
	 Periodic table 	Metals/non-		 Radiation in and out 		Disciplinary knowledge
Key ideas		metals	Physics module 4-	of atoms	Chemistry module 4	Disciplinary
and	Physics module P5		waves		• Group 1	terminology
sequence	Energy	structure	 Wavelength and 	Biology module B4-	• Group 7	 Graph skills Maths skills for
of learning	 Energy transfers 	 Forming ions 	frequency	community level	• Group 0	
Of learning		• Ionic	 Wave properties 	systems	 Reactivity of metals 	science
	Physics module P1	compounds	 Wave speed 			Osmosis investigation analysis to apply skills
	-Matter	• Simple		• Ecosystems	Biology module 5 –	analysis to apply skills
	 Energy, heat & 	molecules		Introduction to	Genes, inheritance and	
	temperature	Giant covalent	Biology module B2-	Sampling	selection	
	Specific heat	 Polymer 	Scaling up	Abiotic and Biotic	• Meiosis	
	capacity	Structure of	Cell differentiation	Factors	Dominant and	
	State changes	metals	Stem cells	Competition and	recessive alleles	
	and specific	 Allotropes of 	 Exchange and 	Interdependence	Genetic crosses	
	latent heat	carbon	transport	Analysis of	 Mutations 	
		 Bulk properties 	Circulatory system	predator/prey graphs	 Natural selection 	

	Biology module B1 - cell level systems Maths skills for science Plant and animal cells Bacterial Cells Light Microscopes Electron Microscopy DNA Structure Enzymes & enzyme reactions Biological Molecules Aerobic respiration	 Changing state Biology module B1 Anaerobic respiration Photosynthesis Factors affecting photosynthesis Limiting factors of photosynthesis Biology module B2- scaling up Diffusion Osmosis Active transport Mitosis 	 Heart and blood Plant transport systems. Transpiration stream Factors affecting transpiration 	 Nutrient cycling Carbon cycle Biology module 5 – Genese, inheritance and selection Variation Sexual and asexual reproduction 	 Evidence of evolution Classification systems Revision for end of year exams 	
Key Questions	 How are particles arranged in solids, liquids and gases? What in inside the atoms? How did we discover the atom? What are the energy stores and how is energy transferred? What happens when we heat solids, liquid and gases? 	 How do gases exert pressure? How are chemical bonds formed? What are the allotropes of carbon? What is the structure of metals and plastics? What is the difference between aerobic and anaerobic respiration? 	 How do we calculate and measure speed? How do we represent motion graphically? What are the properties of waves? How do cells differentiate? How does body cells multiply? What is the importance of stem cells? How does the body transport oxygen? How do plants transport water and 	 What is the EM spectrum and how does it affect our lives? What is ionising radiation and how do we measure it? How do ecosystems interact? How are nutrients recycled through the environment? How does human reproduction 	 How do we separate chemical compounds? How can we determine masses of reactants? How can we purify and analyse chemicals? How can we predict trends in the periodic table? How are our characteristics inherited? What evidence is there for evolution 	 What rules apply to circuits What is the correct terminology to use in science practical work How do we use Maths skills in science

	6. What is inside a cell?7. What happens inside a cell8. How do we digest food?9. How do cells release energy?	 6. How do plants produce and store food? 7. Which factors have an impact on the rate of photosynthesis? 8. Which processes help to transport substances across membranes? 	what factors affect it?	produce variation?	7. How do we organise the natural world?	
Vocabulary	ChemistrySolidLiquidGas	PhysicsDensityPressure	PhysicsVelocityAccelerationDeceleration	PhysicsElectromagneticRadioMicrowave	ChemistryRf valuefiltrationFractional	PhysicsCircuitCurrentPotential difference
	Melting PointBoiling PointIonIsotope	ChemistryRelativeFormula MassGroupPeriod	 Distance Displacement Time m/s km/s 	InfraredVisibleUltravioletX-RayGamma	 Distillation Reactivity Trends Outer electron Displacement 	VoltageAmmeterVoltmeterResistance
	Physics Module Thermal Gravitational Chemical Magnetic Kinetic Nuclear Electrostatic Forces Waves Heating	 lonic Compound Covalent Compound Dot And Cross Diagram Giant Lattice Allotrope Polymer Metals Lattice 	 Equation m/s² Distance-time graph Transverse longitudinal Wavelength Frequency Biology Specialised Acrosome 	 Alpha Beta Half life Ionisation Penetration Attraction Repulsion Unstable isotope Biology Abiotic Factors 	reaction Biology Alleles Asexual reproduction Cancer Clone Continuous variation Diploid cells Discontinuous	Disciplinary knowledge Precision Accuracy Variables Valid Reproducible Repeatable
	• Current		Biconcave	Biomass	Variation	

Specific Heat	Giant	Haemoglobin	Biotic Factors	Dominant allele
Capacity	Covalent	Ciliated		
	Covalent	- 11 11	,	Environmental Variation
· ·	Biology		Community	
Heat	Anaerobic	• Platelets	Competition	
Piology		Meristem	• Consumers	• Gametes
Biology	Respiration	 Differentiation 	 Decomposers 	Sex Cells
Cell membrane		 Meristems 	 Detritivores 	Genetic cross
Cell wall	Oxygen debt	 Mitosis 	 Ecosystem 	Genetic variation
• Chloroplast	• ATP	 Alveoli 	 Egestion 	Genome
• Cytoplasm	• Exothermic	 Artery 	Excretion	Genotype
 Eukaryotic cell 	 Fermentation 	 Atria 	Habitat	Haploid Cells
Flagellum	 Chlorophyll 	 Capillary 	 Interdependence 	Heterozygous
Mitochondria	 Endothermic 	 Diffusion distance 	 Mutualism 	Homozygous
 Nucleus 		 Double Circulatory 	 Parasitism 	Meiosis
 Plasmid 	 Photosynthesis 	System	 Population 	Mutation
 Prokaryotic cell 	 Light intensity 	 Exchange Surface 	 Predation 	Phenotype
 Resolution 	 Active 	Heart	 Producers 	Punnett square
 Magnification 	Transport	 Lumen 	 Discontinuous 	Recessive allele
Electron	 Concentration 	 Xylem 	Variation	Sexual reproduction
microscope	gradient	 Phloem 	Dominant allele	Variation
 Vacuole 	 Diffusion 	 Translocation 	 Environmental 	Zygote
 Active site 	 Osmosis 	 Vascular Bundle 	Variation	• Alleles
 Bases 	 Concentration 	 Potometer 	 Fertilisation 	Asexual
 Chromosome 	gradient	• Vein	 Gametes 	reproduction
 Complementary 	 Water potential 	 Ventricle 	Sex Cells	Cancer
Base pairing	 Turgid 	 Transpiration 		Clone
 Denature 	 Plasmolysis 	•		Continuous variation
• DNA	 Flaccid 			Diploid cells
DNA base	• Lysis			Discontinuous
 Polymer 	 Crenated 			variation
Monomer	• Cell			
Enzyme	• Cyle			
Active site	 DNA replication 			
• Gene	 Mitosis 			
Metabolic rate	 Chromosomes 			
Aerobic	•			
Respiration				

Practical Skills	 Chemistry Comparing chemical and physical changes Specific heat capacity Biology Light Microscopy Investigating Enzyme activity Food Tests 	Physics Density Biology Testing for Starch in plants Investigate factors affecting photosynthesis Investigate osmosis	Physics	Physics	Chemistry Filtration and crystallisation Chromatography	Physics • Building simple circuits using Phet demos Biology • osmosis
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